

Reinhold Environmental Ltd.



2008 APC Round Table
& Expo Presentation

July 13-15, 2008, in Savannah, GA

Good Afternoon

Thank You for the Invitation to Speak Today

Please take a moment to give your pagers
and cell phones a rest; either shut them
off or switch to vibrate mode

Now grab a coffee, a pen/pencil and a
buddy to keep you awake



"Fuels Quality's Effects on Air Pollution Control Devices"

2008 APC Conference

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Hyatt Regency-Savannah
Savannah, GA

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Objective

To present a brief tutorial
on fuel quality's affects on
air pollution control devices

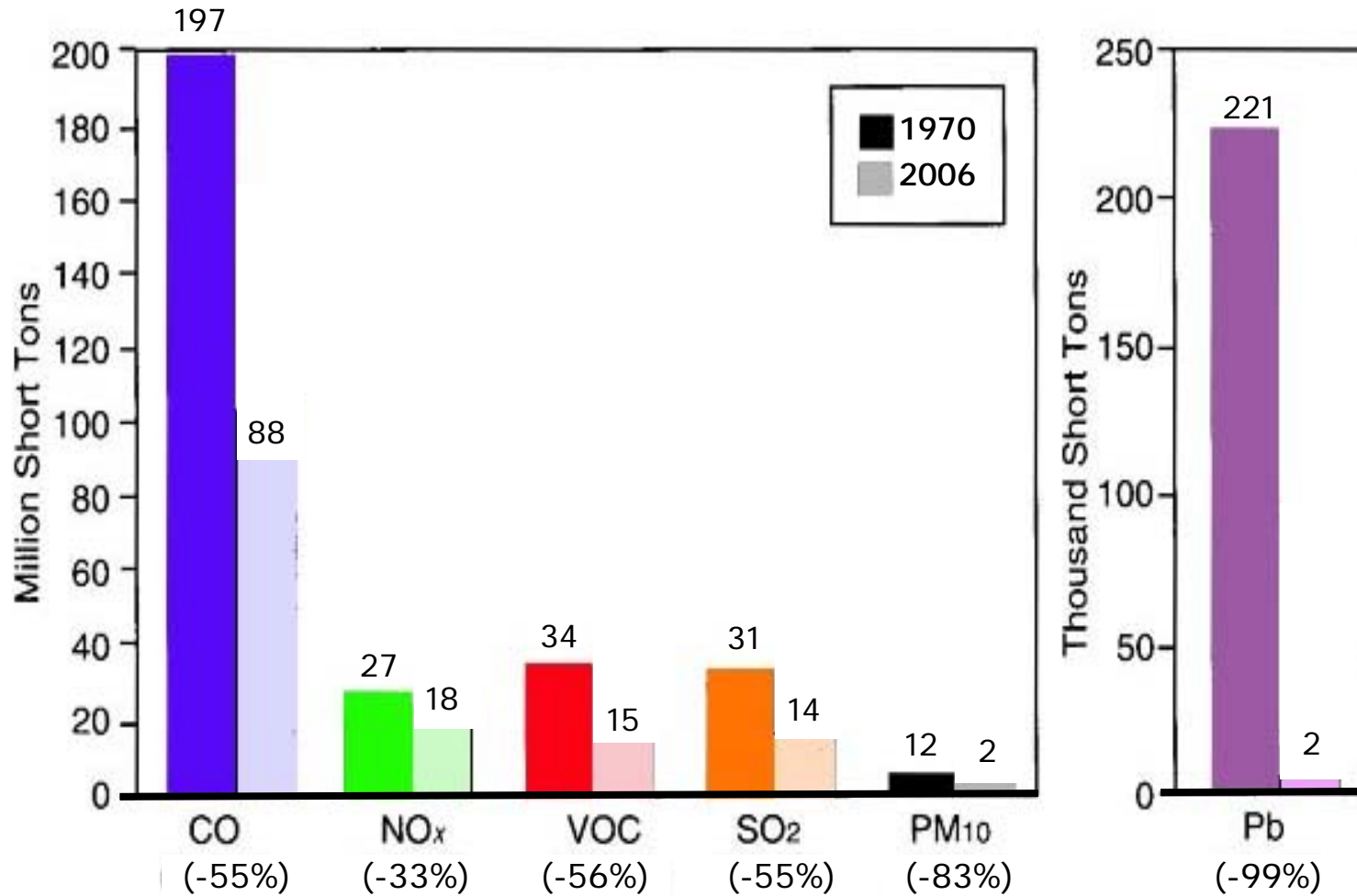


OK, we all think coal is this
nasty fuel that should be
replaced by wind, solar,
biomass, natural gas, etc... &
not to mention nuclear

(Shh... don't say this too loud, but nuclear plants do
not emit greenhouse gases)



Fossil-Fired Emissions 1970-2006 – Criteria Air Pollutants



Comparison of 1970 & 2006 emission of criteria air pollutants, Office of Air Quality Planning & Standards, US EPA

SE Michigan's baseload at 4 AM,
Sunday during Spring & Fall is
>4,000 MWs

So...

What are our energy options
today, Monday, 14 July 2008?



United State's electric power baseload energy mix is nominally:

- Coal 51%
- Nuclear 20%
- Natural Gas 18%
- Hydro 7%
- Oil 3%
- Other 1%

With the exception of nuclear, hydro & some of the Other category, they all emit GHGs and some type of undesirable emissions...NO_x, SO₂, particulate, CO, CO₂, H₂O, HCl, HF, Hg, other miscellaneous HAPs



Since coal is the single largest producer of electricity, and given the time allotted, I'll concentrate on conventional coal combustion

To understand emissions from coal combustion, we must have a basic understanding of:

- coal
- coal combustion
- combustion product deposition
- combustion product collection
 - emissions



What is coal?

**It's definitely not just
black and burns**



Coal Analysis

Mine/Supplier: **Pittsburgh Seam** Sampling Method: **Mechanical**
Sample #: **99F-00852** Sample Date: **2/25/1999**
Sample Description: **S> 1.7%** Sample Receipt: **2/26/1999**

Proximate	<u>As=Received</u>	<u>Dry</u>	<u>MAF</u>
Moisture (-)	4.52		
Ash (-)	6.78	7.10	
Volatile (+)	34.31	35.95	38.68
Fixed Carbon (+)	54.39	56.99	61.32
Sulfur	2.05	2.14	
Heating Value	13,376	14,010	15,080

ADL	2.30
Residual Moisture	2.27

Moisture Load	3.38
Ash Load	5.07
S02	3.07
V/FC	0.63

Ultimate	<u>As=Received</u>	<u>Dry</u>	<u>MAF</u>
Carbon	75.23	78.82	84.84
Hydrogen	5.00	5.24	5.64
Nitrogen	1.45	1.52	1.64
Oxygen	4.97	5.21	5.61

Hardgrove Grindability Index

HGI 50.8 @2.89% Moisture

"Fuels" – 2008 APC-Savannah

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Ash Fusion - 8 point

	<u>Reducing</u>	<u>Oxidizing</u>
Initial	2,209	2,535
Spherical	2,280	2,575
Hemispherical	2,372	2,595
Fluid	2,460	2,665
Plastic Range	251	130

Ash Mineral

SiO2	50.52
Al2O3	23.10
TiO2	0.93
Fe2O3	15.63
CaO	2.64
MgO	0.78
K2O	1.94
Na2O	0.56
S03	1.25
P2O5	0.35
BaO	0.26
SrO	0.81
MnO2	0.01
Undetermined	1.22

B/A	0.29
Ash Ratio	0.22
Ash Type	Bituminous
Slagging Index	1
Slagging Type	Medium
Fouling Index	0.16
Fouling Type	Low
Silica Ratio	0.73
Si/Al	2.19
Fe/Ca	5.92
Dolomite %	15.87

Trace Elements

<u>Trace Elements</u>	<u>Units</u>	<u>Value</u>		<u>Units</u>	<u>Value</u>		<u>Units</u>	<u>Value</u>
Sb, Antimony	ppm	0.11	Pb, Lead	ppm	8.24	Ag, Silver	ppm	0.067
As, Arsenic	ppm	11.97	Mn, Manganese	ppm	18.9	Tl, Thallium	ppm	0.62
Ba, Barium	ppm	104	Hg, Mercury	ppm	0.082	V, Vanadium	ppm	30.1
Be, Beryllium	ppm	2.2	Mo, Molybdenum	ppm	1.12	Zn, Zinc	ppm	14.5
Cd, Cadmium	ppm	0.061	Ni, Nickel	ppm	13.61			
Cl, Chlorine	ppm	957.28	Se, Selenium	ppm	3.22			
Cr, Chromium	ppm	12.4						
Co, Cobalt	ppm	6.21						
Cu, Copper	ppm	15.8						
F, Fluorine	ppm	<10						



Definitions of Coal

- A brown or black combustible sedimentary rock (in the geological sense) composed principally of consolidated and chemically altered plant remains. (ASTM D121)
- Coal is a rock, a sediment, a conglomerate, a biological fossil, a complex colloidal system, an enigma in solid-state physics and an intriguing object for chemical and physical analyses
(van Krevelen)
- Coal is a chemically and physically heterogeneous, "combustible," sedimentary rock consisting of both organic and inorganic material (Bruce Miller, Penn State)



Coalification Process

Materials	Partial Processes	Main Chemical Reactions		C	H	O
Decaying Vegetation ↓ <i>Peat</i> ↓ <i>Lignite</i> ↓ <i>Bituminous coal</i> ↓ <i>Semianthracite</i> ↓ <i>Anthracite</i>	Peatification	Bacterial and fungal life cycles	Wood	49	7	44
	Lignitification	Air oxidation, followed by decarboxylation and dehydration	Peat	60	6	34
	Bituminization	Decarboxylation and hydrogen disproportioning	Lignite	70	5	25
	Preanthracitization	Condensation to small aromatic ring systems	Subbituminous	75	5	20
	Anthracitization	Condensation of small aromatic ring systems to larger ones; dehydrogenation	Bituminous	85	5	10
	Graphitization	Complete carbonification	Anthracite	94	3	3

Increasing Aromatization & Loss of Oxygen with Time

Wt%, dry mmf basis



FIGURE 1-2. The coalification process. (From Van Krevelen, D. W., *Coal: Typology-Physics-Chemistry-Constitution*, Third ed., Elsevier Science, Amsterdam, 1993. With permission).